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Ksp Problems □ Chemistry Name: _____ 1) The value of Ksp of AgCl is 1.8×10^{-10} . What would be the molar concentration of Ag⁺ and Cl⁻ in pure water placed in contact with solid AgCl(s)? $K_{sp} = [Ag^+][Cl^-]$ $1.8 \times 10^{-10} = [x][x]$ $x = 0.000013416$ M or 1.3416×10^{-5} M 2) The molar solubility of PbCl ...

Ksp Problems □ Chemistry

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3□3 Practice Problems 1. How many protons and electrons are 8. What is the name of element that has present in a vanadium atom? atoms that contain 25 protons? 2. How many protons and electrons are 9. What is the name of the element that has present in a nitrogen atom? atoms that contain 82 protons? 3. How many protons and electrons are 10.

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0.710 kg of sodium carbonate (washing soda), Na_2CO_3 , in 10.0 kg of water a saturated solution at 0°C ; 125 g of NH_4NO_3 in 275 g of water a mixture used to make an instant ice pack; 25 g of Cl_2 in 125 g of dichloromethane, CH_2Cl_2 ; 0.372 g of histamine, $\text{C}_5\text{H}_9\text{N}$, in 125 g of chloroform, CHCl_3

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