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Acid: $\text{HCO}_3^- \text{ (aq)} + \text{OH}^-$?

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$\text{CO}_3^{2-}(\text{aq}) + \text{H}_2\text{O}(\text{l})$
 Base: $\text{H}_3\text{O}^+(\text{aq}) + \text{HCO}_3^-(\text{aq})$
 $\text{H}_2\text{O}(\text{l}) + \text{H}_2\text{CO}_3(\text{aq})$
 Weak acids 108. $\text{H}_2\text{O}(\text{l}) + \text{HB}(\text{aq}) \rightleftharpoons \text{H}_3\text{O}^+(\text{aq}) + \text{B}^-(\text{aq})$
 $K_a = \frac{[\text{H}_3\text{O}^+(\text{aq})][\text{B}^-(\text{aq})]}{[\text{HB}(\text{aq})]}$
 $3.500 = 3.16 \times 10^{-4} \text{ mol/L}$

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 $2 [\text{HB}(\text{aq})] (0.370 \text{ mol/L} -$
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Acid = HCl , base = H_2O 3

Acid = HCOOH , base = KOH 4

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Acid = HCl , base = CH_3COOH
5 Acid = HCl , base = NH_3
Acid = HCO_3^- , base = OH^-

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Acid Base Salt Identifying

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Acids (pages 240–241) 1.

Define an acid. Match these
common acids to their uses.

Acids Uses 2. acetic acid 3.

sulfuric acid 4.

hydrochloric acid 5. carbonic

acid 6. nitric acid 7.

Describe some general

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Properties of acids. 8.

Place the following substances in the correct column in the table: lemons, vinegar, grapefruit, sour milk, tomatoes.

Chapter 8: Solutions, Acids,

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Worksheet. 4. 5.1 Reading
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Acid and Base Worksheet -

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Answers. 1) Using your knowledge of the Brønsted-Lowry theory of acids and bases, write equations for the following acid-base reactions and indicate each conjugate acid-base pair: a) $\text{HNO}_3 + \text{OH}^- \rightarrow \text{H}_2\text{O} + \text{NO}_3^-$.

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HNO_3 and NO_3^- make one pair
 OH^- and H_2O make the other.

b) $\text{CH}_3\text{NH}_2 + \text{H}_2\text{O} \rightleftharpoons \text{CH}_3\text{NH}_3^+ + \text{OH}^-$

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Y
Magnesium oxide (MgO),
calcium carbonate (CaCO_3)
and sodium hydrogen
carbonate (NaHCO_3). A strong
acid or base completely
ionises in solution.

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reaction = reaction

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proton e.g. $\text{HCl} + \text{NaOH} \rightarrow$
 $\text{NaCl} + \text{H}_2\text{O}$ H^+ donates H^+ + ?
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this type of question you
have to know the approximate
pKa values discussed in
class and have an idea of

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what they mean in terms of relative acid strength and also relative base strength.
a.

3719 Acids and Bases

Worksheet - Peter Norris YSU

a reaction of an acid with a

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You, yielding a solution that is not as acidic or basic as the starting solutions were. pH scale. a range of values used to express the concentration of hydrogen ions in a solution. hydroxide ion. a positively

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You charged ion formed of a
hydrogen atom that has lost
its electron.

chapter three lesson 3-5
acids, bases, and solutions

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Brønsted-Lowry. The

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with Mg, Zn to form $H_2(g)$
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