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Chapter 9 Review Stoichiometry Section 2 Work CHAPTER 9 REVIEW Stoichiometry SECTION 2 PROBLEMS Write the answer on the line to the left Show all your work in the space provided 1 45 mol The following equation represents a laboratory preparation for oxygen gas: 2KClO₃(s) → 2KCl(s) + 3O₂(g) How many moles of O₂ form if 30 mol of

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qualified staff provide an efficient and personal customer service. Chapter 9 Stoichiometry Worksheet Mole review Worksheet. Calculate the molar mass of the following, and then write the molar mass in ratio form. KNO₃ . Na₂SO₄ . Ca(OH)₂ (NH₄)₂SO₃. CuSO₄. Moles and Mass. Determine the ...

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and one molecule of oxygen gas. Macroscopic: Two moles of hydrogen peroxide (present in ...

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Chapter 9 Stoichiometry Section 2 Worksheet SECTION 2 continued Date Class _____ 602 9 421 1 a \ tt mash 01 ox aen Cas i pridued it 100 of lithium c a C ti l o c i o g di l C1O c — LCi(,; — h The oxygen gas produced in part ahas density of 143 g/L aiculate the olurne of Chapter 9 - Stoichiometry 9-2 Ideal Stoichiometric Calculations Ideal Stoichiometry - All reactants are converted into ...

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Antoine Lavoisier's great accomplishments include the discovery of oxygen's role in combustion, helping to develop the metric system, writing the first extensive list of elements, helping to reform the nomenclature of chemistry, and the discovery that while matter may change shape through chemical reaction its mass remains the same. It is for these extraordinary accomplishments that he is often referred to as the "Father of Modern Chemistry." Some scholars argue that this moniker is more the result of self-promotion and that his discoveries relied heavily on the work of others, nonetheless his impact on advancing this field of science cannot be understated. "Elements of Chemistry" was first published in 1790 and is largely concerned with the chemistry of combustion. While modern students of chemistry might find the work limited in its scope, the historical impact of its publication cannot be understated. The experiments contained within helped to lay the foundation for the understanding of the role of oxygen, hydrogen, acids, and alcohols in chemical reactions and its emphasis on quantitative analysis and instrumentation helped to establish the use of chemistry as a legitimate science for understanding and defining the physical world.

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